

Water plays a fundamental role in life and, although a tremendous amount of research has been performed on this seemingly simple substance, there are still many unresolved questions. The literature is virtually unlimited and it is impossible to cover more than a few topics in this book. I have chosen to concentrate on snow and ice, and I hope the book will whet the appetite to learn more about this fascinating field. The selected topics are mostly rather briefly treated but further information is easily available on the Internet. It therefore seems unnecessary to include a long list of references here.

Water plays a unique role in chemistry. The special properties of the different forms of water — from ice and snow to liquid water — are due to hydrogen bonding between the H₂O molecules, and this book is a tribute to a field to which I have spent a major part of my research. The hydrogen bond is of fundamental importance in biological systems since all living matter has evolved from and exists in an aqueous environment. Hydrogen bonds are involved in most biological processes as little energy is needed in forming as well as breaking these bonds.

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